Sustainable Thermoplastics from Renewable Resources: Thermal behavior of Poly(1,4-cyclohexane dimethylene 2,5-furandicarboxylate)

Nejib Kasmi¹, **Niki Poulopoulou²**, Zoi Terzopoulou¹, Dimitrios G. Papageorgiou³, Dimitrios N. Bikiaris¹, George Z. Papageorgiou²

¹Laboratory of Polymer Chemistry and Technology, Department of Chemistry, Aristotle University of Thessaloniki, GR-541 24, Thessaloniki, Macedonia, Greece

²Chemistry Department, University of Ioannina, P.O. Box 1186, 45110 Ioannina, Greece

³School of Materials and National Graphene Institute, University of Manchester, Oxford Road, Manchester M13 9PL, UK

Telephone: +30 2651 008354, fax :+30 2651 008795 and e-mail address: nikki_p @windowslive.com

Indicate if POSTER X or ORAL ☐ presentation

SCHIPLA BEPAUL

Abstract

Poly(1,4-cyclohexane dimethylene 2,5-furandicarboxylate) (PCHDMF) а sustainable thermoplastic that can be prepared from renewable resources with potential uses as a replacement for its terephthalate (PCHDMT) and naphthalate (PCHDMN) homologues. A detailed study was undertaken to assess its thermal behavior and chemical/structural characteristics, in comparison to its counterparts. The melting temperature of PCHDMF was observed at $T_m=264.5$ °C, the glass transition was obtained at 77°C and the cold crystallization temperature was seen at 121°C. The melting of the polymers was studied under a variety of conditions and all samples displayed the characteristic melting-recrystallization-remelting behavior. Isothermal and dynamic crystallization tests revealed that PCHDMF crystallizes at faster rates than its homologues, while the equilibrium melting point of PCHDMF was established at 300 °C. The enthalpy of fusion values for the polyesters were found $\Delta H_{\rm m}$ =137 J/g for PCHDMF, $\Delta H_{\rm m}$ =108 J/g for PCHDMT, $\Delta H_{\rm m}$ =119 J/g for PCHDMN. Using the Lauritzen-Hoffman analysis of spherulite growth rates, larger K_g values were found for PCHDMF, due to its less flexible structure.

References

[1] Papageorgiou, GZ., Tsanaktsis, V., Bikiaris DN.: Synthesis of poly (ethylene furandicarboxylate) polyester using monomers derived from renewable resources.

thermal behavior comparison with PET and PEN. Phys. Chem. Chem. Phys. 16(17),7946-7958 (2014).

[2] Papageorgiou, GZ., Papageorgiou, DG., Terzopoulou, Z., Bikiaris, DN.: Production of bio-based 2, 5-furan dicarboxylate polyesters: Recent progress and critical aspects in their synthesis and thermal properties. European Polymer Journal. 83, 202-229 (2016).

